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Key Pdf Element

Symbol Atomic .

Number Mass.

Number Number

of. neutrons

nitrogen-15 8

Beryllium-9 4 3.

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information to determine the atomic mass of chlorine. Two isotopes are known: chlorine-35 (mass = 34.97 amu) and chlorine-37 (mass = 36.97 amu). The relative abundance's are

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change to Dalton's
atomic theory is
that atoms are
divisible into
subatomic
particles: Electrons,
protons, and
neutrons are
examples of these

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fundamental particles There are many other types of particles, but we will study these three ...

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The Structure of the Atom Figure 4.2 In his book A

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Chemical
Philosophy, John
Dalton presented
his symbols for the
elements known at
that time and their
possible
combinations. John
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the concept of the
atom was revived
in the eighteenth
century, it took

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Key Measuring
Atomic Mass
Instead of grams,
the unit we use is
the Atomic Mass
Unit (amu) It is

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defined as one-twelfth the mass of a carbon-12 atom. Carbon-12 chosen because of its isotope purity. Each isotope has its own atomic mass, thus we determine the average from percent abundance.

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